ST. JOSEPH'S COLLEGE OF ARTS & SCIENCE (AUTONOMOUS) CUDDALORE-1



PG & RESEARCH DEPARTMENT OF CHEMISTRY

M.Phil - SYLLABUS 2020-2021
M.Phil. CHEMISTRY

CURRICULUM DESIGN TEMPLATE FROM 2021- 2022

			PG	AND RE	SEARCH DE	PARTMENT OF CHEMISTRY			
					CURRICUL	UM TEMPLATE			
					M.Phil	. Chemistry			
					SEM	ESTER – I			
S.No	S.No Part Hours/ Week Credit Course Code Course Title							ximum]	1
		<u></u>	WCCK		Coue		CIA	ESE	TOTAL
1	III	Core Theory - 1	7	5	MPCH101	Research Methodology	25	75	100
2	III	Core Theory - 2	7	5	MPCH102	Advanced Chemistry	25	75	100
3	III	Core Theory - 3	7	5		Elective Paper (Guide Paper)	25	75	100
4	III	Library	12	-		Science-6 (Library)+6(Lab)	-	-	-
S	emes	ter Total	33	15			75	225	300
					SEMI	ESTER – II			
S.No		Part	Hours/	Credit	Course	Course Title	Ma	ximum]	Marks
3.110		rart	Week	Credit	Code	Course Title	CIA	ESE	TOTAL
5	III	Core Theory - 1		21	JCH201	Dissertation and Viva Voce	100	100	200
S	emes	ter Total		21			100	100	200
Total Credits				36					

Syllabus

M.Phil (CH)		MPCH101
SEMESTER - I	RESEARCH METHODOLOGY	HRS/WK – 7
CORE – I		CREDIT- 5

Objective:

To impart knowledge on research methodology. To gain an in depth knowledge in statistical analysis.

COURSE OUTCOMES (COs)

CO1: Scholars learn the meaning, objective and problems in Research.

CO2: Scholars acquire the basic principles of experimental designs.

CO3: Scholars get to know about Data collection methods for documentation and presentation.

CO4: Scholars learn data analysis, types and sources of errors and determination of control errors.

CO5: Scholars learn essentials of a scientific report.

Relationship Matrix Course Outcomes, Programme Outcomes and Programme Specific Outcomes

SEMESTER I	(COUI	RSE	COD	E:			CO	URSI	ETITI	LE:			HOURS:	CREDITS:
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]	PRO	GRA	MM	E		P	ROGI	RAMN	1E SP	ECIFI	C		MEAN S	CORE OF
COURSE				ES(P						AES(P					O'S
OUTCOMES	D()1	DO3	DO3	DO4	DO5	DSA1	DSO2	DCU3	DSOA	DSO5	DSA	DSO7	DSAG		
	101	1 02	1 03	104	103	1 301	1 302	1 303	1 504	1 303	1300	1307	1 500		
CO1	3	4	3	4	3	4	4	4	4	4	4	4	3	3.	69
CO2	3	3	3	3	3	3	3	3	4	3	3	4	4	3.	23
CO3	3	3	4	3	3	3	3	4	4	4	4	4	4	3.	53
CO4	3	3	3	3	3	3	3	4	4	3	4	4	4	3.	38
CO5	3	3	3	4	3	3	3	3	3	4	4	4	4	3.	38
			Mea	n Ov	erall S	Score								3.	44

Result: The Score of this Course is 3.44 (High)

Association	1%-20%	21%-40%	41%-60%	61%-80%	81%-100%
Scale	1	2	3	4	5
Interval	0<=rating<=1	1.1<=rating<=2	2.1<=rating<=3	3.1<=rating<=4	4.1<=rating<=5
Rating	Very Poor	Poor	Moderate	High	Very High

This Course is having **High** association with Programme Outcome and Programme Specific Outcome.

UNIT-1 RESEARCH METHODOLOGY

[12 Hrs]

Meaning of research – the objective of research – motivation of research – approaches, and significance – methods versus methodology – research in scientific methods – research process – criteria for good research – problem encounters by research in India – funding agencies.

UNIT - II: RESEARCH DESIGN

[12 Hrs]

Research problem: selecting the problem – the necessity of defining the problem – techniques involved in defining the problem – research design – needs and features of good design – different research design – basic principles of experimental designs.

UNIT – III: DATA COLLECTION AND DOCUMENTATION

[12 Hrs]

Data collection methods – data types – processing and presentation of data- techniques of ordering data – the meaning of primary and secondary data – the uses of computers in research – the library and internet – uses of search engines – virtual libraries – common software for documentation and presentation.

UNIT - IV: DATA AND ERROR ANALYSIS

[12 Hrs]

Statistical analysis of data – standard deviation – the correlation – comparison of sets of data – chi-squared analysis for data – characteristics of probability distribution – binomial, Poisson and normal distribution – the principle of least square fittings – curve fitting – a measurement of errors – types and sources of errors – determination of control errors.

UNIT - V: RESEARCH COMMUNICATION

[12 Hrs]

Meaning of research report – logical format for writing and paper – essential of a scientific report: abstract- introduction, review of the literature – materials and methods and discussion – write up steps in drafting report – effective illustrations: tables and figures – reference styles: Harvard and Vancouver systems.

Text Books:

- 1. Research Methodology, methods, and techniques-C.R.Kothari-Wishwa Prakasam publications, II Edition.
- 2. Research: An Introduction-Robert Ross-Harper and Row Publications.
- 3. Research methodology-P.Saravanavel-Kitlab Mahal, Sixth edition.
- 4. A Hand-Book of Methodology of Research-Rajammal P.A.Devadass-Vidyalaya press.
- 5. N.Subramanian, Introduction to Computer.

Reference Books:

- 1. G.W.Secdecor and W.Cocharan, Statistical methods oxford and IBH, New Delhi.
- 2. Santosh Gupta, Research methodology methods, and statistical techniques.
- 3. S.P.Gupta, Statistical Methods-
- 4. Scientific social surveys and research-P.Young-Asia publishers, Bombay.
- 5. How to write and publish a scientific paper –R.A. Day Cambridge University Press.
- 6. Thesis and assignment writing-Anderson-Wiley Eastern Ltd.

M.Phil (CH)		MPCH102
SEMESTER – I	ADVANCED CHEMISTRY	HRS/WK – 7
CORE – II		CREDIT- 5

Objective:

To study the applications of spectroscopy and to apply it in practice. To provide hands on experience in instrumental methods.

COURSE OUTCOMES (COs)

CO1: Scholars learn the instrumental methods of GC- HPLC, CV, Polarography and Amperometry.

CO2: Scholars understand the Principles and applications in structural elucidation.

CO3: Scholars learn the Applications of UV-Visible, IR, NMR in Organic molecules.

CO4: Scholars learn the Applications of UV-Visible, IR, NMR, Mossbauer and ESR spectrometry in the determination of structures of inorganic molecules.

CO5: Scholars learn the concept of point groups and retro synthesis.

Relationship Matrix Course Outcomes, Programme Outcomes and Programme Specific Outcomes

SEMESTER I	(COUI			E:		A 1			E TITI		X 7		HOURS:	CREDITS:
			PCH					DVAN						/	3
		PRO	GRA	MM	\mathbf{E}		P	ROGE	RAMN	1E SP	ECIFI	C		MEAN S	CORE OF
COURSE		UTC				OUTCOMES(PSO)									O'S
OUTCOMES	PO1	PO2	PO3	PO4	PO5	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6	PSO7	PSO8		
CO1	4	4	3	4	3	4	4	4	4	4	4	4	3		76
CO2	4	3	3	3	3	3	3	3	4	3	3	4	4	3.	30
CO3	3	3	4	3	3	3	3	4	4	4	4	4	4	3.	53
CO4	4	3	3	3	3	3	3	4	4	3	4	4	4	3.	46
CO5	3	3	3	4	3	3	3	3	3	4	4	4	4	3.	38
			Mea	n Ov	erall S	Score								3.	48

Result: The Score of this Course is 3.48 (High)

Association	1%-20%	21%-40%	41%-60%	61%-80%	81%-100%
Scale	1	2	3	4	5
Interval	0<=rating<=1	1.1<=rating<=2	2.1<=rating<=3	3.1<=rating<=4	4.1<=rating<=5
Rating	Very Poor	Poor	Moderate	High	Very High

This Course is having **High** association with Programme Outcome and Programme Specific Outcome.

UNIT – I: INSTRUMENTAL METHODS OF ANALYSIS

[12 Hrs]

Atomic absorption and emission spectroscopy, chromatography: GC - HPLC, electroanalytical methods: coulometry cyclic voltammetry, polarography, amperometry, and ion-selective electrodes.

UNIT - II: SPECTROSCOPY

[12 Hrs]

Principles and applications in structural elucidation

Rotational – diatomic molecules – isotopic substitution and rotational constants. Vibrational – diatomic molecules – linear triatomic molecules – specific frequencies of functional groups in polyatomic molecules. Electronic – singlet and triplet states – np* and pp*transitions – application to conjugated double bonds and conjugated carbonyls – Woodward-Fieser rules – charge transfer spectra.nuclear magnetic resonance – basic principle – chemical shift – spin-spin interaction and coupling constant. Mass spectroscopy – parent peak, base peak – metastable peak – MCLafferty rearrangement.

UNIT – III [12 Hrs]

Applications of UV-Visible, IR, NMR – COSY, NOESY, HMBC, HSQC and mass spectrometry in the determination of structures of organic molecules.

UNIT – IV [12 Hrs]

Applications of UV-Visible, IR, NMR, Mossbauer and ESR spectrometry in the determination of structures of inorganic molecules – a variation of optical activity with wavelength – optical rotatory dispersion and circular dichroism curves and their application in determining the configuration and conformation of different inorganic compounds and conformational analysis.

UNIT – V [12 Hrs]

Symmetry elements – point groups – optical activity – its origin – atomic and conformation asymmetry – a variation of optical activity with wavelength. Retrosynthesis – synthons – synthetic equivalents – GI – target molecules – retrosynthesis of molecules (cubane, ciprofloxacin)

Text Books:

- 1. H.H.Willand, L.L. Merrit and J.A.Dean, Instrumental Methods of Analysis-D.Ven. Nostrand& Co.
- 2. H.A. Strobel, Chemical Instrumentation, Addition-Wesley publishing & Co.
- 3. R.S.Drago, Physical Methods in Inorganic Chemistry
- 4. R.S.Drago, Physical Methods in Chemistry.

Reference Books:

- 1. C.N.Banwell, Fundamentals of Molecular Spectroscopy, 1996, McGraw Hill.
- 2. William Kemp, Organic Spectroscopy, Macmillan Ltd, 1994.
- 3. R.M.Silverstein, G.C.Basler, and T.C.Morril Spectrometric Identification of Organic Compounds, John Wiley-1997.
- 4. Stuart Warren -Designing Organic Synthesis

Question paper pattern for M.Phil

THEORY EXAMINATION

Internal Examination (25 marks)	Internal	l Examination		(25)	marl	ks))
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Two Internal Examinations 15 marks
Assignment / Seminar 10 marks
Total 25 marks

External Examination (75 marks)

Question Pattern

M. Phil. CHEMISTRY

Time: 3 Hours Max. Marks: 75

Section A $(5 \times 6 = 30 \text{ marks})$

ANSWER ALL FIVE QUESTIONS

Internal Choice (Either or Pattern)

Section B $(3 \times 15 = 45 \text{ marks})$

ANSWER ANY THREE QUESTIONS

Out of Six Questions (Open Choice)

TOTAL (30+45=75)

NOTE: Equal weightage will be given for all units.

M.Phil (CH)	DISSEDTATION AND VIVA VOCE	JCH201
SEMESTER – II	DISSERTATION AND VIVA-VOCE	
PROJECT	For the students admitted in the year 2009	CREDIT- 21

Objective:

The main objective of the project is to expose the students to research and industrial atmosphere and to get a broad idea to develop project.

COURSE OUTCOMES (COs)

CO1: Ability to perform critical thinking, reasoning and creative thinking.

CO2: Ability to use the technology.

CO3: Ability to visualize the problems and provide solutions.

CO4: Ability to test technical skills.

CO5: Ability to work both independently and in groups on development of projects.

Relationship Matrix Course Outcomes, Programme Outcomes and Programme Specific Outcomes

SEMESTER	(COURSE CODE:					COURSE TITLE:								CREDITS:
II						DISSERTATION AND VIVA-VOCE									21
		PRO	GRA	MM	E	PROGRAMME SPECIFIC								MEAN S	CORE OF
COURSE	C)UT(COM	ES(P	PO)	OUTCOMES(PSO)							CO	O'S	
OUTCOMES	PO1	PO2	PO3	PO4	PO5	PSO1	PSO2	PSO3	PSO4	PSO5	PSO6	PSO7	PSO8		
CO1	5	4	5	5	4	4	4	4	4	3	4	4	4	4.	10
CO2	5	4	5	5	4	4	4	4	5	3	4	4	4	4.	20
CO3	5	5	5	5	5	5	5	4	5	3	4	4	4	4.	50
CO4	5	5	5	5	5	5	5	4	5	3	4	4	4	4.	50
CO5	5	5	5	5	5	5	5	4	5	3	4	4	4	4.	50
			Mea	n Ov	erall S	Score								4.	36

Result: The Score of this Course is 4.36 (Very High)

Association	1%-20%	21%-40%	41%-60%	61%-80%	81%-100%
Scale	1	2	3	4	5
Interval	0<=rating<=1	1.1<=rating<=2	2.1<=rating<=3	3.1<=rating<=4	4.1<=rating<=5
Rating	Very Poor	Poor	Moderate	High	Very High

This Course is having **Very High** association with Programme Outcome and Programme Specific Outcome.

FORMAT FOR PREPARING THE PROJECT DISSERTATION REPORT

Arrangement of Contents

- 1. Title Page
- 2. Bonafide Certificate
- 3. Acknowledgement
- 4. Table of contents
- 5. Abstract
- 6. Chapters of the Report
- 7. References
- 8. Appendices, if any

Appendices should be named as APPENDIX - A

APPENDIX - B

BINDING SPECIFICATION

Report should be found using flexible cove of thick white art paper. The Spine for the bound volume should be of black of 2cms width. The Cover should be printed in block letters.

MARGIN SPECIFICATION

Top : 1.25 inches
Bottom : 1.25 inches
Left : 1.50 inches
Right : 1.25 inches

PAGE NUMBERING

All page numbers should be typed without punctuation on the bottom-center position of the page. The Preliminary pages (table of contents and abstract) should be numbered in lowercase roman literals. Papers of main text, starting with chapter-1, should be consecutively numbered using Arabic numerals.

TITLE PAGE

<TITLE OF THE PROJECT>

A DISSERTATION REPORT SUBMITTED TO
ST. JOSEPH'S COLLEGE OF ARTS & SCIENCE (AUTONOMOUS)
IN PARTIAL FULFILLMENT
OF THE REQUIREMENT FOR THE AWARD OF
THE DEGREE OF

MASTER OF PHILOSOPHY IN CHEMISTRY

BY

<NAME OF THE STUDENT> (REGISTER NO: X00XXX00)

Under the Guidance of

<NAME OF THE RESEARCH GUIDE>

<Designation & Department>



PG & RESEARCH DEPARTMENT OF CHEMISTRY ST.JOSEPH'S COLLEGE OF ARTS & SCIENCE (AUTONOMOUS) CUDDALORE – 607001

<MONTH & YEAR>

CERTIFICATE

<NAME OF THE RESEARCH GUIDE>

<Designation>
PG & Research Departm

PG & Research Department of Chemistry, St.Joseph's College of Arts & Science (Autonomous),

Cuddalore - 607001.

CERTIFICATE

This is to certify that this Project Dissertation report entitled, "<TITLE OF THE PROJECT DISSERTATION>" is a bonafide record of work done by <NAME OF THE STUDENT>, (<ROLL NUMBER>) under my supervision and submitted to ST.JOSEPH'S COLLEGE OF ARTS & SCIENCE (AUTONOMOUS), CUDDALORE – 1, Affiliated to ANNAMALAI UNIVERSITY, ANNAMALAI NAGAR in partial fulfillment for the award of the Degree of MASTER OF PHILOSOPHY IN CHEMISTRY.

Head of the Department		Research Guide
		Principal
Submitted for the Viva-Voce I	Examination held on	
Examiners:	_	
1.	2.	

SCHEME OF EVALUATION

DISSERTATION AND VIVA-VOCE

Continuous Internal Assessment (CIA) (100 marks)

Based on the Periodical Evaluation of Record, Reviews and Experiments, Assessed by the Research Guide.

External Examination (100 marks)

Based on the evaluation of the Project Dissertation Report submitted and the Viva-Voce examination, Assessed by the External Examiner.